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P2-

- 1. What is scientific method? What are the various parts of the scientific method? (10 points)
- 2. Identify each of the following as a pure substance, a homogeneous mixture, or a heterogeneous mixture. (10 points)
 - a. gasoline

- d. brass
- b. a stream with gravel at the bottom
- e. copper metal

- c. air
- 3. Name each binary compound. (10 points)
 - a. CsF b. AlCl₃ c. MgI₂
- 4. Under appropriate conditions at 1000°C, ammonia gas reacts with oxygen gas to produce gaseous nitrogen monoxide (common name, nitric oxide) and gaseous water. Write the unbalanced and balanced equations for this reaction. (10 points)
- 5. When the prediction was made for whether a reaction will occur, what are the "driving forces" that pull the reactants toward products. (10 points)
- 6. Isopentyl acetate, $C_7H_{14}O_2$, the compound responsible for the scent of bananas, can be produced commercially. Interestingly, bees release about 1 μg (1x10⁻⁶ g) of this compound when they sting. This attracts other bees, which then join the attack. How many moles and how many molecules of isopentyl acetate are released in a typical bee sting? (10 points)

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P2-2

7. Propane, C₃H₈, when used as a fuel, reacts with oxygen to produce carbon dioxide and water according to the following unbalanced equation:

$$C_3H_8(g) + O_2(g) \rightarrow CO_2(g) + H_2O(g)$$

What mass of oxygen will be required to react exactly with 96.1g of propane? (10 points)

- 8. What is ideal gas law? (10 points)
- 9. Although milk is not a true solution (it is usually a suspension of tiny globules of fat, protein, and other substrates in water), it does contain a dissolved sugar called lactose. Cow's milk typically contains 4.5% by mass of lactose, C₁₂H₂₂O₁₁. Calculate the mass of lactose present in 175g milk. (10 points)
- 10. What is buffered solution? And how to use pH meter properly. (10 points)